

# Al-Be-Sc (Aluminum-Beryllium-Scandium)

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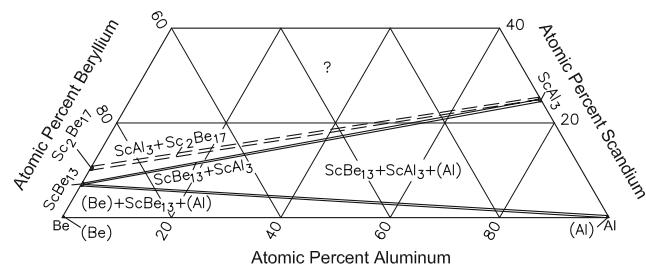
Recently, [2003Fri] determined an isothermal section for this ternary system at 600 °C in the Sc-lean region, which depicts no ternary phases.

## Binary Systems

The Al-Be phase diagram [Massalski2] is of the simple eutectic type, with very little terminal solid solubility between Al and Be. The Al-Sc phase diagram [Massalski2] depicts the following intermediate compounds:  $\text{ScAl}_3$  ( $L_1$   $\text{AuCu}_3$ -type cubic),  $\text{ScAl}_2$  ( $C15$ ,  $\text{MgCu}_2$ -type cubic),  $\text{ScAl}$  ( $B2$ ,  $\text{CsCl}$ -type cubic) and  $\text{Sc}_2\text{Al}$  ( $B8_2$ ,  $\text{Ni}_2\text{In}$ -type hexagonal). Using 15 binary alloys, [2003Fri] determined the Be-Sc phase diagram. It depicts only the three known Be-Sc compounds:  $\text{ScBe}_{13}$  ( $D2_3$ ,  $\text{NaZn}_{13}$ -type cubic),  $\text{Sc}_2\text{Be}_{17}$  ( $\text{Th}_2\text{Ni}_{17}$ -type or  $\text{Th}_2\text{Zn}_{17}$ -type), and  $\text{ScBe}_5$  ( $D2_d$ ,  $\text{CaCu}_5$ -type hexagonal). The eutectic between (Be) and  $\text{ScBe}_{13}$  is at 1280 °C and between 0.1 and 0.15 at.% Sc.  $\text{ScBe}_{13}$  and  $\text{ScBe}_5$  form congruently, whereas  $\text{Sc}_2\text{Be}_{17}$  forms through a peritectic reaction [2003Fri].

## Ternary Phase Equilibria

With starting metals of 99.99% Al, 99.6% Be, and 99.6 or 99.975% Sc, [2003Fri] induction melted under Ar atm about 20 ternary alloys, containing up to 17.5 at.% Sc. The alloys were annealed at 600 °C for 800 h and quenched in water. The phase equilibria were studied by metallography, x-ray



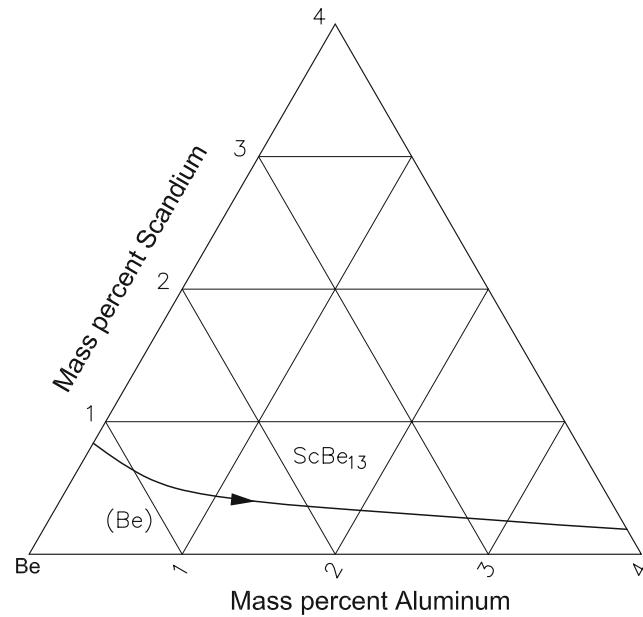
**Fig. 1** Al-Be-Sc isothermal section at 600 °C for Sc-lean alloys [2003Fri]

powder diffraction, electron probe microanalysis, differential thermal analysis and hardness measurements. The isothermal section at 600 °C for the Sc-lean region constructed by [2003Fri] is shown in Fig. 1. The region is characterized by the three-phase fields of (Be) +  $\text{ScBe}_{13}$  + (Al), (Al) +  $\text{ScBe}_{13}$  +  $\text{ScAl}_3$ , and  $\text{ScBe}_{13}$  +  $\text{Sc}_2\text{Be}_{17}$  +  $\text{ScAl}_3$ . The solubility of the third component in the binary compounds is negligible. No ternary phases were found.

The partial results obtained by [2003Fri] on the solidification features indicate that the monovariant liquidus line originating from the eutectic point on the Be-Sc side slopes down towards the Be-Al side, as shown in Fig. 2.

## Reference

**2003Fri:** I.N. Fridlyander and L.V. Molchanova, Interaction of Beryllium with Aluminum and Scandium, *Metally*, (5), p 109-114, in Russian; TR: *Russ. Metall.*, 2003, (5), p 476-480



**Fig. 2** Al-Be-Sc monovariant liquidus line for L + (Be) +  $\text{ScBe}_{13}$  equilibrium [2003Fri]